

Math 220 – Section 3.2 Solutions

1. The general equation we use for mixing problems is:

$$\frac{dx}{dt} = r_i c_i - r_o \frac{x}{V_0 + (r_i - r_o)t}$$

In this problem,

$$r_i = r_o = 8 \frac{\text{L}}{\text{min}}, \quad c_i = 0.05 \frac{\text{kg}}{\text{L}}, \quad V_0 = 100 \text{ L}$$

After simplifying, the differential equation is:

$$\frac{dx}{dt} + 0.08x = 0.4$$

This equation is first order, linear, and separable. Therefore, we can use two methods to solve it. Let's compute the integrating factor:

$$\mu(t) = \exp\left(\int 0.08 dt\right) = \exp(0.08t) = e^{0.08t}$$

The solution is:

$$\begin{aligned} x(t) &= \frac{1}{\mu(t)} \left[\int \mu(t)Q(t) dt + C \right] \\ x(t) &= e^{-0.08t} \left[\int 0.4e^{0.08t} dt + C \right] \\ x(t) &= 5 + Ce^{-0.08t} \end{aligned}$$

The initial condition is $x(0) = 0.5$ kg. Using this to solve for C we get:

$$\begin{aligned} 0.5 &= 5 + Ce^{-0.08(0)} \\ C &= -4.5 \end{aligned}$$

Therefore, the mass of salt as a function of time is:

$$x(t) = 5 - 4.5e^{-0.08t}$$

The concentration of salt is:

$$c = \frac{\text{mass}}{\text{volume}}$$

To determine when $c = 0.02$ kg/L, we must solve the following equation for t :

$$\begin{aligned} 0.02 &= \frac{x(t)}{V(t)} \\ 0.02 &= \frac{5 - 4.5e^{-0.08t}}{100} \end{aligned}$$

Using some algebra, we find that:

$$t = -\frac{\ln \frac{3}{4.5}}{0.08} = 5.07 \text{ min}$$

3. This was done in class. Check your notes for the solution.

4. The general equation we use for mixing problems is:

$$\frac{dx}{dt} = r_i c_i - r_o \frac{x}{V_0 + (r_i - r_o)t}$$

In this problem,

$$r_i = 4 \frac{\text{L}}{\text{min}}, r_o = 3 \frac{\text{L}}{\text{min}}, c_i = 0.2 \frac{\text{kg}}{\text{L}}, V_0 = 100 \text{ L}$$

After simplifying, the differential equation is:

$$\frac{dx}{dt} + \frac{3}{100+t}x = 0.8$$

This equation is first order and linear. Let's compute the integrating factor:

$$\mu(t) = \exp\left(\int \frac{3}{100+t} dt\right) = \exp(3 \ln(100+t)) = (100+t)^3$$

The solution is:

$$\begin{aligned}x(t) &= \frac{1}{\mu(t)} \left[\int \mu(t)Q(t) dt + C \right] \\x(t) &= (100+t)^{-3} \left[\int 0.8(100+t)^3 dt + C \right] \\x(t) &= (100+t)^{-3} [0.2(100+t)^4 + C] \\x(t) &= 0.2(100+t) + C(100+t)^{-3}\end{aligned}$$

The initial condition is $x(0) = 0$ kg since the tank initially contained pure water. Using this to solve for C we get:

$$\begin{aligned}0 &= 0.2(100+0) + C(100+0)^{-3} \\0 &= 20 + \frac{C}{100^3} \\C &= -20(100)^3\end{aligned}$$

Therefore, the mass of salt as a function of time is:

$$x(t) = 0.2(100+t) - 20 \left(\frac{100}{100+t} \right)^3$$

The concentration of salt is:

$$c = \frac{\text{mass}}{\text{volume}}$$

To determine when $c = 0.1$ kg/L, we must solve the following equation for t :

$$\begin{aligned}0.1 &= \frac{x(t)}{V(t)} \\0.1V(t) &= x(t) \\0.1(100+t) &= 0.2(100+t) - 20 \left(\frac{100}{100+t} \right)^3\end{aligned}$$

After some algebra, we find that:

$$t = 100(\sqrt[4]{2} - 1) = 18.92 \text{ min}$$